TECHNICAL DATA SHEET

UREA

Reduced Biuret Urea (RBU)

Properties					SDS #1132	
Total Nitrogen % by weight (guaranteed)			46.0 minimum			
Water % by weight 0.2						
Biuret % by weight maximum (guaranteed)				0.4		
pH by weight 8.5 – 9.5						
Bulk Density lbs/cubic foot 46						
Fertilizer Nutrient Designation			46 - 0 - 0			
Typical Size Distribution*						
Screen	-6+8	-8+10	-10+12	-12+14	-14	
Average % Retained	6	55	37	1	1	
Cumulative %	6	61	98	99	100	
*Tyler Mesh Screen Analysis						

Hazardous Shipping Description

- There are no DOT restrictions, other than weight, to transport UREA prill.
- A large spill of UREA should be recovered dry. All attempts should be made to keep it from dissolving into a vegetated drainage. The high nitrogen content (46%) can kill foliage if not diluted. Dissolved UREA can be handled, if neccessary, by a municipal water treatment facility.
- Consult SDS #1132 for more specific and comprehensive information about chemical hazards.

PRODUCT DESCRIPTION

The REDUCED BIURET UREA prill is a small diameter,

spherical white solid. It is an organic molecule called an amide containing 46% nitrogen in the form of amine groups. This UREA is specially manufactured by quickly melting UREA crystal and prilling. Unlike traditional prilling, the short heating step from crystal to prill results in a very low BIURET formation. UREA is infinitely soluble in water and is a benign and safe chemical to handle.

APPLICATION RECOMMENDATIONS

- The REDUCED BIURET UREA is used specifically as a fertilizer for leaf crops that are sensitive to BIURET. UREA prill may be used as a slow release fertilizer. It must be decomposed by microorganisms before it can be assimilated by plants.
- ALWAYS exercise caution when using this chemical as fertilizer because it has the highest nitrogen content of any solid.

TRANSPORTATION, STORAGE AND HANDLING

- UREA will decompose into ammonia and carbon dioxide at 275°F.
- ALWAYS wash vessels containing UREA thoroughly before attempting repairs requiring welding.
- NEVER allow UREA to come into contact with nitric acid. The resulting chemical is unstable and dangerous.

ADDITIONAL INFORMATION – Visit <u>dynonobel.com</u> for Brochures and Case Studies related to this product.

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